On the CH Stretching Bands of Surface Alcoholates Formed on Metal Oxides—A Reply to Morrow, Thomson and Wetmore

In their letter, Morrow, Thomson and Wetmore (1) discussed the assignments of the CH stretching bands of surface alcoholates. They stated that since the degeneracy of an asymmetric vch mode is removed in the methyl group attached to oxygen, a single asymmetric $\nu_{\rm CH}$ mode of the methyl group is no longer preserved in surface methoxide. This was experimentally illustrated by the spectrum of surface methoxide formed on silica. From the spectra of methoxy CH₃O- and dideuterated methoxy CHD₂O- groups, they concluded that while the band due to a symmetric v_{CH} mode could be assigned, those due to asymmetric $\nu_{\rm CH}$ mode was somewhat more complicated. The spectrum of methoxy group CH₃O- observed by Morrow, Thomson and Wetmore (1) was practically the same as those observed by other workers (2, 3). However, in the experiments performed by Morrow, Thomson and Wetmore excess methanol was evacuated at room temperature so that physically adsorbed methanol may be present together with methoxide. This may complicate the spectra. Actually, Borello, Zecchina and Morterra (2) reported that methanol strongly and physically adsorbed on silica could be removed by evacuation at temperatures over 200°C.

In spite of these complexities of the spectra, however, it is reasonable to conclude that a single asymmetric $\nu_{\rm CH}$ mode does not always exist in surface methoxide as Morrow, Thomson and Wetmore (1) noted.

In the spectra of bulk methoxides, generally three strong bands are seen which are referred to as ν_a , ν_b and ν_c bands, respectively, in the decreasing order of wave number. The exceptions are aluminum and germanium methoxides in 12 different bulk methoxides. The ν_a band was generally assigned to the overtone of the CH bending Copyright © 1973 by Academic Press, Inc.

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mode whereas v_b and v_c bands were asymmetric and symmetric modes, respectively. As Morrow, Thomson and Wetmore suggested, a more detailed investigation should be carried out for the assignments of the bands.

The surface methoxides so far obtained on silica (2,3) and germania (4) exhibited three strong bands in the $\nu_{\rm CH}$ region and the ν_b band was found to be most intense. When methanol was brought in contact with magnesia or calcium oxide at room temperature, methoxide was formed.* The features of the spectra were similar to those observed on silica and germania (7). With reference to the work by Morrow, Thomson and Wetmore (1), Borello, Zecchina and Morterra (2) and Low and Harano (3), the ν_c band can be assigned to a symmetric v_{CH} mode although v_a and v_b bands can not be assigned in a strict sense. For surface boron (3), zinc (5) and aluminum methoxides (6), two strong bands were observed. The bands located at a higher wave number were stronger in the former two methoxides hence they could be assigned to the v_b band and to the ν_c band, respectively. As for the latter methoxide, we could not estimate the intensities of the two bands from the spectrum in the literature (6) but from the correlation we previously reported (8) they could also be assigned to v_b and v_c bands. Figure 1 shows the positions of strong $\nu_{\rm CH}$ bands of surface methoxides observed by several authors. The most intense band is marked with a thick line. Similar plots are

*Tench, Giles and Kibblewhite [Trans. Faraday Soc. 67, 854 (1971)] recently reported that the $\nu_{\rm CR}$ bands of methoxide on magnesia were observed at 2920, 2860 and 2807 cm⁻¹ in excellent accord with our results.

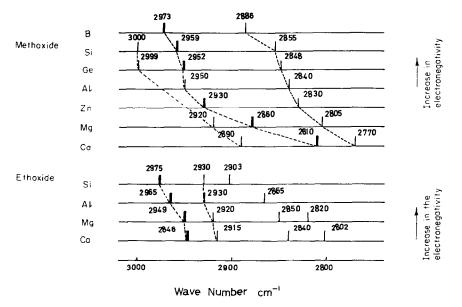


Fig. 1. The CH stretching bands of surface alcoholates formed on metal oxides. The band with highest intensity represents with a thick line. Some results are from the literature: methoxides on SiO₂ and B in SiO₂ (3); methoxide on GeO₂ (4); methoxide on ZnO (5); methoxide on Al₂O₃ (6); methoxides on MgO and CaO, and ethoxides on SiO₂ and CaO (8); ethoxide on Al₂O₃ (9) and ethoxide on MgO (10). Ethoxide was formed on magnesium oxide at room temperature. The feature of the spectrum was similar to that observed by Kagel and Greenler [J. Chem. Phys. 49, 1638 (1968)] although the positions of the bands in rcH region were not given by these authors.

also shown for surface ethoxides. As shown in Fig. 1, it is apparent that all strong bands in the $\nu_{\rm CH}$ region shift in the same direction when metal oxide differs. In this respect, even though no single asymmetric $\nu_{\rm CH}$ mode is preserved, the positions of the bands in the $\nu_{\rm CH}$ region can be well correlated with the electronegativity of metal in metal oxides. As was discussed in the previous work (8), it is suggested that on a less electronegative oxide surface alkoxy-group is anionic whereas covalently bound alkoxides are formed on a more electronegative oxide.

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